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The first enantioselective synthesis of (-)-microbiotol and (+)- β -microbiotene

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Abstract—The first enantioselective total synthesis of (–)-microbiotol and (+)- β -microbiotene, sesquiterpenes containing three neighboring quaternary carbon atoms belonging to the cyclocuparane group, starting from cyclogeraniol employing a Sharpless–Katsuki asymmetric epoxidation, a boron trifluoride etherate mediated epoxide rearrangement and an intramolecular diazo ketone cyclopropanation as key steps, is described.

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1. Introduction

The cyclocuparane group is a small class of tricyclic sesquiterpenes, containing a 4-methyl-1-(1,2,2-trimethylcyclopentyl)bicyclo[3.1.0]hexane **1** carbon framework comprising of three contiguous quaternary carbon atoms. Even though the first two members of this class of sesquiterpenes, grimaldone **2** and microbiotol **3** were isolated in 1975 and 1981, their structures were only elucidated in 1988 and 1991, respectively.^{1,2} Isolation of cyclocuparanol **4** was first reported in 1984 by Asakawa et al.³ from *Marchantia polymorpha* and *Marchantia paleacea* Bertol. var *diptera* (Mont) Hatt., and was the first one to be established to contain the cyclocuparane carbon framework. Isolation of microbiotol **3** was first reported by Raldugin et al. in 1981, from the ether extract of the needles of *Microbiota decussata*, an evergreen bush, which grows on the Sikhote–Alin mountain ridge, while the structure was established by Trachev et al.² in 1991 on the basis of spectral data and molecular mechanics calculations in conjunction with conformational analysis by NMR. The absolute configuration of microbiotol 3 was established via its chemical transformation to (R)-cuparene (+)-5. It is worth noting that microbiotol 3, isolated from the plant source is antipodal to cyclocuparanol 4 and grimaldone 2, isolated from the liverworts. In 1998, Konig et al.⁴ reported the isolation of three more cyclocuparane sesquiterpenes, α -microbiotene 6, β -microbiotene 7, and isogrimaldone 8, in addition to grimaldone 2 from the essential oil of Mannia fragrance. The presence of a 1,2,2-trimethylcyclopentyl substituted bicyclo[3.1.0]hexane carbon framework containing three contiguous quaternary carbon atoms made the cyclocuparanes challenging synthetic targets.⁵ Herein, we report the first enantioselective total synthesis of (-)-microbiotol 3 and (+)- β -microbiotene 7.



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2. Results and discussion

Recently, we have reported^{5b} the synthesis of racemic cyclocuparanes 3, 4, and 7 by employing the tricyclic ketone 9a as the key precursor, which was obtained from Hagemann's ester via intramolecular cyclopropanation of the diazo ketone derived from acid 10. Hence, a methodology was conceived for the enantioselective generation of ester 11 starting from the readily available cyclogeraniol 12.

The sequence is depicted in Scheme 1. Katsuki–Sharpless asymmetric epoxidation of cyclogeraniol 12 using (+)-diethyl tartrate following the reported procedure⁶ generated the (1*S*,6*S*)-epoxide (-)-13, $[\alpha]_D^{25} = -26.5$ (*c* 1.1, CHCl₃), with an enantiomeric excess of 95%. Oxidation of the primary alcohol in 13 with phosphorus pentoxide–dimethyl sulfoxide and triethylamine⁷ furnished the epoxyaldehyde 14, $[\alpha]_D^{25} = +48.3$ (*c* 1.5, CHCl₃), in 48% yield. Horner–Wadsworth–Emmons reaction with triethyl phosphonoacetate and sodium hydride transformed aldehyde 14 into the epoxy propenoate 15 in 77% yield, whose structure was established from its spectral data.¹¹ For the ring contraction of the six-membered ring in 15, a Lewis acid mediated rearrangement of epoxide was adopted. Thus, treatment of epoxy ester 15 with boron trifluoride etherate in methylene chloride

at -70 °C for 1 h cleanly furnished ketoester 16 in 90% yield.¹¹ The configuration of the stereogenic center in 16 was assigned as R in analogy to the rearrangement of epoxides derived from β -methylcarvone,⁸ and finally confirmed by its conversion to (-)-microbiotol 3. However, the enantiopurity of 16 has been slightly reduced $(87\%)^9$ in comparison to geraniol epoxide 13 as the reaction is not stereospecific and the intermediacy of a carbonium ion in the rearrangement. Hydrogenation of the olefin in ketoester 16 with 10% palladium over carbon as the catalyst in ethanol furnished keto ester 17 in a quantitative yield.¹¹ Since a conventional Wittig reaction was not successful, ketoester 17 was converted into pentenoate 11 employing the procedure developed by Lombardo.¹⁰ Thus, reaction of ketoester 17 with the reagent prepared from methylene bromide, zinc, and titanium tetrachloride furnished pentenoate 11 in 60% yield.¹¹ Pentenoate 11 was transformed into microbiotol and β -microbiotene by employing a previously reported^{5b} sequence. Thus, hydrolysis of ester 11 with 5% sodium hydroxide in 1:1 methanol-water furnished acid 10, which was converted into diazo ketone 18 via the corresponding acid chloride. Anhydrous copper sulfate-copper catalyzed intramolecular cyclopropanation of diazo ketone 18 furnished a 1.7:1 mixture of norcyclocuparanones **9a**, $[\alpha]_{D}^{25} = -25$ (*c* 0.4, CHCl₃) and **9b**, $[\alpha]_{D}^{25} = -11.4$ (*c* 0.7, CHCl₃), which were separated by



Scheme 1. Reagents, conditions, and yields: (a) Ref. 6; (b) DMSO, P_2O_5 , 0 °C, 30 min; Et₃N, 1 h; 48%; (c) (EtO)₂P(O)CH₂COOEt, NaH, THF, 8 h, 77%; (d) BF₃:Et₂O, CH₂Cl₂, -70 °C, 1 h, 90%; (e) H₂ (1 atm), 10% Pd/C, EtOH, 12 h, 99%; (f) TiCl₄, CH₂Br₂, Zn, CH₂Cl₂, 0 °C, 1 h, 60%; (g) 5% NaOH, H₂O–MeOH (1:1), reflux, 12 h; (h) (COCl)₂, C₆H₆, rt, 2 h; CH₂N₂, Et₂O, 0 °C, 2 h; (i) CuSO₄, Cu, *c*-C₆H₁₂, reflux, W-lamp, 5 h, 75%, 11; (j) MeMgI, Et₂O, 0 °C, 2 h, 75%; (k) Ph₃P⁺CH₃I⁻, NaO'Am, C₆H₆, rt, 2 h, 75%.

silica gel column chromatography. Structures of norketones **9a** and **9b** were established by comparison of the



spectral data with the racemic compounds.^{5b} Wittig methylenation of ketone **9a** furnished β -microbiotene **7**, $[\alpha]_D^{25} = +7.4$ (*c* 0.4, CHCl₃). On the other hand, addition of methylmagnesium iodide transformed tricyclic ketone **9a** into (–)-microbiotol **3**, $[\alpha]_D^{25} = -11.4$ (*c* 0.8, CHCl₃), lit.² $[\alpha]_D^{25} = -13$ (*c* 3.08, CHCl₃). In a similar manner, addition of methylmagnesium iodide to the tricyclic ketone **9b** furnished epimicrobiotol **19**, $[\alpha]_D^{25} = -24$ (*c* 0.5, CHCl₃).

3. Conclusion

In conclusion, we have accomplished the first enantioselective total synthesis of microbiotol **3** and β -microbiotene **7**, cyclocuparane sesquiterpenes containing three contiguous quaternary carbon atoms, employing a Katsuki–Sharpless asymmetric epoxidation of cyclogeraniol, a Lewis acid catalyzed epoxide rearrangement and an intramolecular cyclopropanation of a diazo ketone as key steps.

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References

- 1. Huneck, S.; Connolly, J. D.; Freer, A. A.; Rycroft, D. S. *Phytochemistry* **1988**, *27*, 1405.
- Trachev, A. V.; Shakirov, M. M.; Raldugin, V. A. J. Nat. Prod. 1991, 54, 849–853.
- Asakawa, Y.; Toyota, M.; Bischler, H.; Campbell, E. O.; Hattori, S. J. Hattori Bot. Lab. 1984, 57, 383; Rycroft, D. S.; Cole, W. J. J. Chem. Res. (S) 1998, 600–601.
- Melching, S.; Blume, A.; Konig, W. A.; Muhle, H. *Phytochemistry* 1998, 48, 661–664.
- (a) Srikrishna, A.; Ramachary, D. B. *Tetrahedron Lett.* 2000, 41, 2231–2232; (b) Srikrishna, A.; Ramachary, D. B. *Tetrahedron Lett.* 1999, 40, 6669–6670; (c) Srikrishna, A.; Nagamani, S. A. J. Chem. Soc., Perkin Trans. 1, 1999, 3393–3394.
- Oritani, Y.; Yamashita, K. Phytochemistry 1983, 22, 1909–1912; Abad, A.; Agullo, C.; Arno, M.; Cunat, A. C.; Zaragoza, R. J. Synlett 1993, 895–896.
- Taber, D. F.; Amedio, J. C., Jr.; Jung, K. Y. J. Org. Chem. 1987, 52, 5621–5622.
- Reaction of the *anti* and *syn*-epoxides of β-methylcarvone with *p*-toluenesulfonic acid generated diketones I and II, respectively, in a highly stereoselective manner (structures of I and II were confirmed via X-ray crystal analysis of a

derivative). Srikrishna, A.; Ramasastry, S. S. V. Unpublished results.



- 9. Based of the conversion of ketoester 16 into natural microbiotol (-)-3 with 87% ee.
- Lombardo, L. *Tetrahedron Lett.* **1982**, *23*, 4293–4296;
 Pine, S. H.; Ed. Paquette, L. A. Org. React. **1993**, *43*, 1.
- 11. Yields refer to the isolated and chromatographically pure compounds. All the compounds exhibited spectral data (IR, ¹H and ¹³C NMR, and mass) consistent with their structures. Selected spectral data for ethyl E-3-(2,2,6trimethyl-7-oxabicyclo[4.1.0]heptan-1-yl)prop-2-enoate 15: $[\alpha]_{D}^{25} = +86.5$ (c 2, CHCl₃). IR (neat): v_{max}/cm^{-1} 1721, 1653. ¹H NMR (300 MHz, CDCl₃ + CCl₄): δ 7.10 (1H, d, J 15.6 Hz), 5.95 (1H, d, J 15.6 Hz), 4.17 (2H, q, J 7.2 Hz), 2.10-1.80 (1H, m), 1.73 (1H, t of d, J 15.0 and 5.7 Hz), 1.55–1.35 (3H, m), 1.31 (3H, t, J 7.2 Hz), 1.53 (6H, s), 1.13–1.10 (1H, m), 0.94 (3H, s). ¹³C NMR (75 MHz, CDCl₃ + CCl₄): δ 165.7 (C), 144.0 (CH), 124.5 (CH), 70.3 (C), 65.3 (C), 60.1 (CH₂), 35.8 (CH₂), 33.6 (C), 30.0 (CH₂), 26.0 (CH₃), 26.1 (CH₃), 20.9 (CH₃), 17.1 (CH₂), 14.5 (CH₃). HRMS: m/z Calcd for C₁₄H₂₂O₃Na (M+Na): 261.1467. Found: 261.1466. For ethyl E-4-oxo-4-(1,2,2trimethylcyclopentyl)but-2-enoate **16**: $[\alpha]_D^{25} = -1.3$ (*c* 1.5, CHCl₃). IR (neat): v_{max}/cm^{-1} 1725, 1689, 1631. ¹H NMR (300 MHz, CDCl₃ + CCl₄): δ 7.38 (1H, d, J 15.5 Hz), 6.65 (1H, d, J 15.5 Hz), 4.25 (2H, q, J 6.9 Hz), 2.52–2.38 (1H, m), 1.85-1.60 (3H, m), 1.60-1.40 (2H, m), 1.34 (3H, t, J 6.9 Hz), 1.20 (3H, s), 1.10 (3H, s), 0.86 (3H, s). ¹³C NMR $(75 \text{ MHz}, \text{ CDCl}_3 + \text{CCl}_4): \delta 202.8 \text{ (C)}, 165.5 \text{ (C)}, 137.6$ (CH), 130.4 (CH), 61.1 (CH₂), 59.2 (C), 44.4 (C), 40.6 (CH₂), 34.4 (CH₂), 25.5 (CH₃), 24.7 (CH₃), 20.3 (CH₃), 19.8 (CH₂), 14.3 (CH₃). Mass: mlz (C₁₄H₂₂O₃) 238 (M⁺, 3), 193 (9), 169 (10), 128 (62), 111 (66). For ethyl 4-oxo-4-(1,2,2-trimethylcyclopentyl)butanoate 17: $[\alpha]_D^{25} = -23.1$ (*c* 1.3, CHCl₃). IR (neat): v_{max}/cm^{-1} 1736, 1699. ¹H NMR (300 MHz, CDCl₃ + CCl₄): δ 4.04 (2H, q, J 7.2 Hz), 2.80-2.50 (2H, m), 2.50-2.25 (3H, m), 1.80-1.30 (5H, m), 1.20 (3H, t, J 7.2 Hz), 1.08 (3H, s), 1.02 (3H, s), 0.78 (3H, s). ¹³C NMR (75 MHz, CDCl₃ + CCl₄): δ 212. 2 (C), 172.4 (C), 60.2 (CH₂), 59.2 (C), 44.3 (C), 40.2 (CH₂), 35.3 (CH₂), 34.7 (CH₂), 28.0 (CH₂), 25.6 (CH₃), 24.5 (CH₃), 21.0 (CH₃), 19.7 (CH₂), 14.4 (CH₃). Mass: m/z 195 (M⁺-OEt, 12), 171 (9), 151 (8), 131 (20), 129 (30), 125 (75), 111 (80). HRMS: m/z Calcd for C₁₄H₂₄O₃Na (M+Na): 263.1623. Found: 263.1620. For ethyl 4-(1,2,2-trimethylcyclopentyl)pent-4-enoate **11**: $[\alpha]_{D}^{25} = -18.6$ (*c* 0.7, CHCl₃). IR (neat): v_{max}/cm^{-1} 1738, 1631, 891. ¹H NMR (300 MHz, CDCl₃ + CCl₄): δ 4.88 (1H, s), 4.75 (1H, s), 4.12 (2H, q, J 7.2 Hz), 2.50-2.30 (4H, m), 2.30-2.10 (1H, m), 1.75-1.55 (4H, m), 1.55–1.40 (1H, m), 1.26 (3H, t, *J* 7.2 Hz), 1.04 (6H, s), 0.80 (3H, s). ¹³C NMR (75 MHz, CDCl₃ + CCl₄): δ 173.1 (C), 153.7 (C), 109.1 (CH₂), 60.2 (CH₂), 51.9 (C), 43.7 (C), 40.4 (CH₂), 36.9 (CH₂), 33.6 (CH₂), 28.5 (CH₂), 26.5 (CH₃), 24.8 (CH₃), 23.0 (CH₃), 19.3 (CH₂), 14.4 (CH₃). Mass: m/z 238 (M⁺, 5), 182 (14), 169 (22), 149 (15), 137 (20), 123 (63), 109 (66), 95 (100). HRMS: *m*/*z* Calcd for C₁₅H₂₆O₂Na (M+Na): 261.1830. Found: 261.1832.